

Part II

Section - I

Q Nos Differentiate between eukaryotic and prokaryotic cell.

Eukaryotic Cell

Prokaryotic Cell

Eu means True
Eukaryotic means
True Nucleus

Pro means false
Prokaryotic means
false Nucleus.

Eukaryotic Cell
have a definite
Nucleus

Prokaryotic Cell
not have definite
Nucleus.

It contain
Linear DNA

It contain
Circular DNA

It have double
membrane or
membrane bound
organelles

It have
no organelles.

Large Ribosome

Small Ribosome.

Multicellular or
few unicellular

Unicellular.

Contain Mitochondria,

It has no

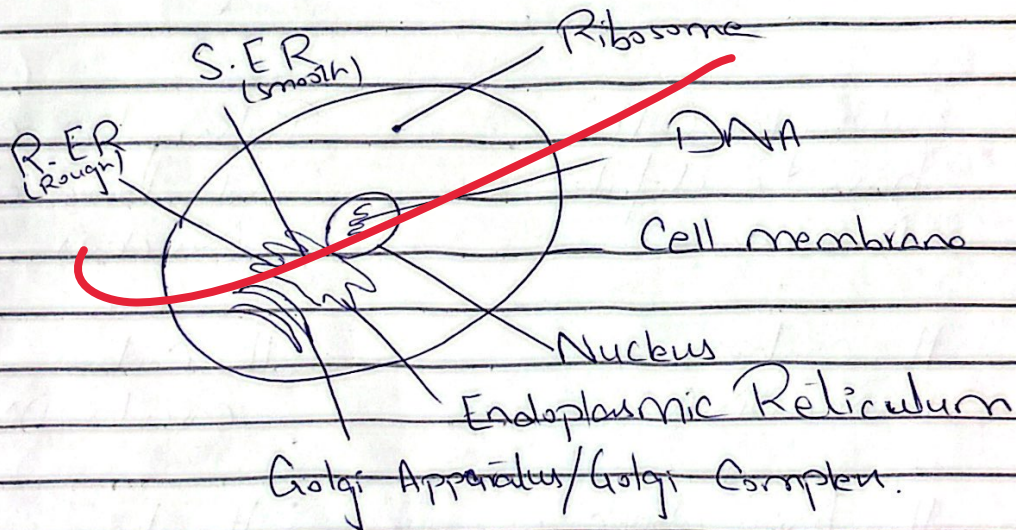
endoplasmic reticulum

no mitochondria
endoplasmic reticulum

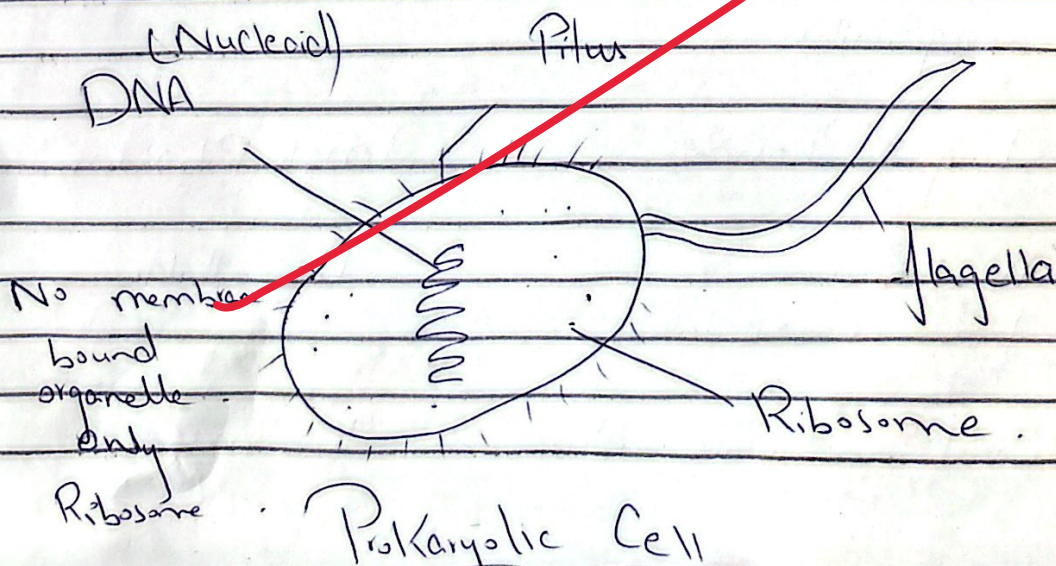
Plants, fungi, animals are some examples of eukaryotic cells

Bacteria and Archaea are the examples of prokaryotic cells

Difference b/w two in the form of their diagrams



Animal Eukaryotic Cell



Prokaryotic Cell

Q No 5 (b) What is Global Warming?
What is Kyoto Protocol?

What is Global Warming -

Global Warming refers to the gradual increase of the temperature of the earth due to the excessive CO_2 emissions along with other pollutants.

Factors that cause Global Warming:-

- Man made factors.
- Natural factors.

Man made factors:-

Deforestation:-

Man excess use of trees in their building materials resulted in deforestation. Trees are the natural source of removing pollutants from the atmosphere. Deforestation alter the natural work.

Use of fertilizers and Pesticides:-

The use of fertilizers for the good production of crops and use of Pesticides for the control of Pests. Both fertilizers and Pesticides result in the pollutant increment.

Use of Vehicles:

The use of vehicles results in various greenhouse emissions. Vehicles burn fossil fuels which emit a large amount of Carbon dioxide and other toxins into the atmosphere resulting in a temperature increase.



gaseous emission

Chlorofluorocarbons:

The use of air conditioners, refrigerators, have been adding CFC's into the environment which affects the atmospheric ozone layers.

Over population:-

The more people breath more is the Carbon dioxide emission and hence global warming.

Natural factors.

Volcanoes.

Volcanoes are one of the largest natural contributors to the global warming.

Forest Blazes: Natural fire erupt in the forests which results in the deforestation and hence global warming.

Effect of Global Warming:-

Climate change

Spread of disease

High Mortality Rates.

Loss of Natural Habitat

Rise in Temperature

Threats to Ecosystem.

Q What is Kyoto Protocol.

Kyoto Protocol is an international agreement signed which enforced countries to take step to control green house gas emissions.

It is applied to the seven green house gases -

Carbon dioxide (CO_2)

Methane (CH_4), Nitrous oxide (N_2O)

Per-fluoro carbons (PFCs)

Hydrofluorocarbons (HFCs)

Sulfur Hexafluoride (SF_6)

Nitrogen Trifluoride (NF_3)

Target of Kyoto Protocol.

The Kyoto Protocol signed in Kyoto, Japan on 11 Dec 1997 and entered into force on 16 Feb 2005 aimed at reducing green house gas emissions by an average of 5% below 1990 levels over the 5 year period from 2008 to 2012.

Mechanism of Kyoto Protocol.

The Kyoto Protocol will work as:

By reducing the emission of green house gases

Plantation

Reforestation

Preventing deforestation

Industrialization using cleaning and renewable energy.

Reasons of failure of Kyoto Protocol:-

It had no compulsion on developing countries.

India and China, the main polluters were out of this agreement as they come under the developing countries.

This placed their competitors USA, Canada and Russia at the great disadvantage and hence USA withdrew itself.

⇒ Achievements of Kyoto Protocol:-

It was a step forward in a right direction.

It led to Paris agreement.

(Q No 5 c) Write a detail note on GIS

GIS:

GIS stands for Geographic Information System. It is a system designed to capture, evaluate, manipulate, handle, and view all forms of geographical and spatial information and data.

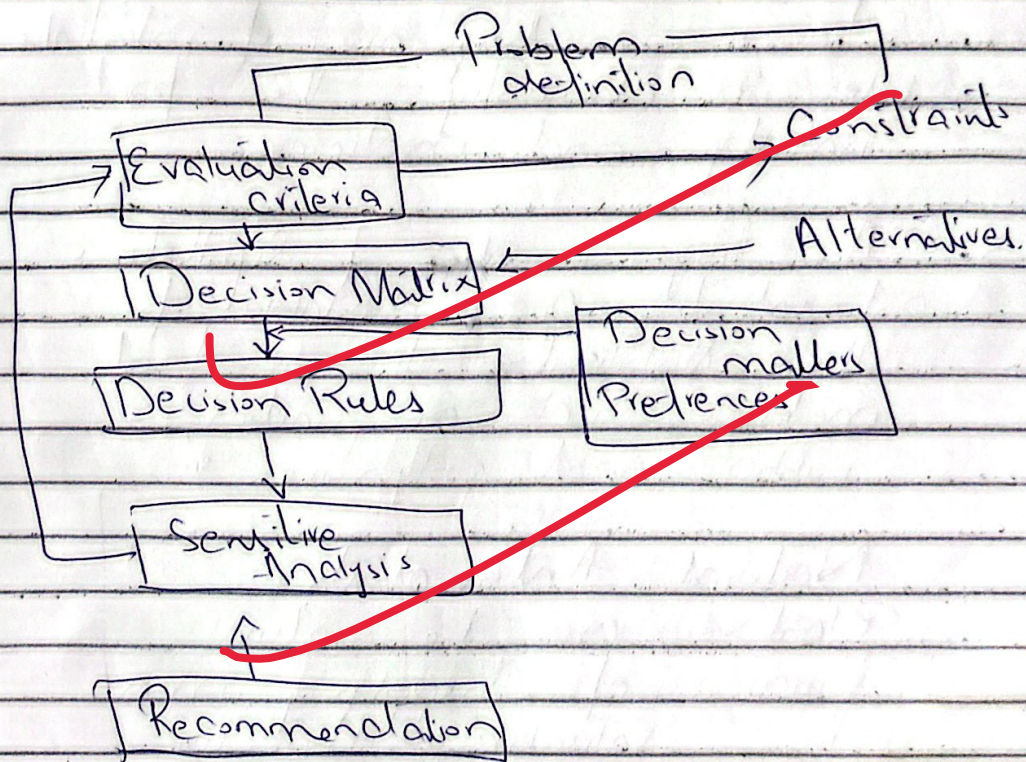
Advantages of GIS:

A GIS stores information on

geographical characteristics and their features.

These characteristics are recognized as representation of points, lines, regions or routers. Road records can be processed as lines, for example in the map of a city and borders could be documented as zones, and aerial photos can be saved as raster data.

⇒ Flow chart of the process steps followed in GIS.



QNo 5 a) Briefly describe Antioxidants

Antioxidants:-

Antioxidants are defined as substances that when present in food, delay, control or inhibit oxidation and deterioration of food quality.

Antioxidants are compounds that inhibit oxidation.

Types of Antioxidants:-

Synthetic Antioxidants

Natural Antioxidants

Synthetic Antioxidants:-

They are of potential use in chemistry, the food industry and medicine.

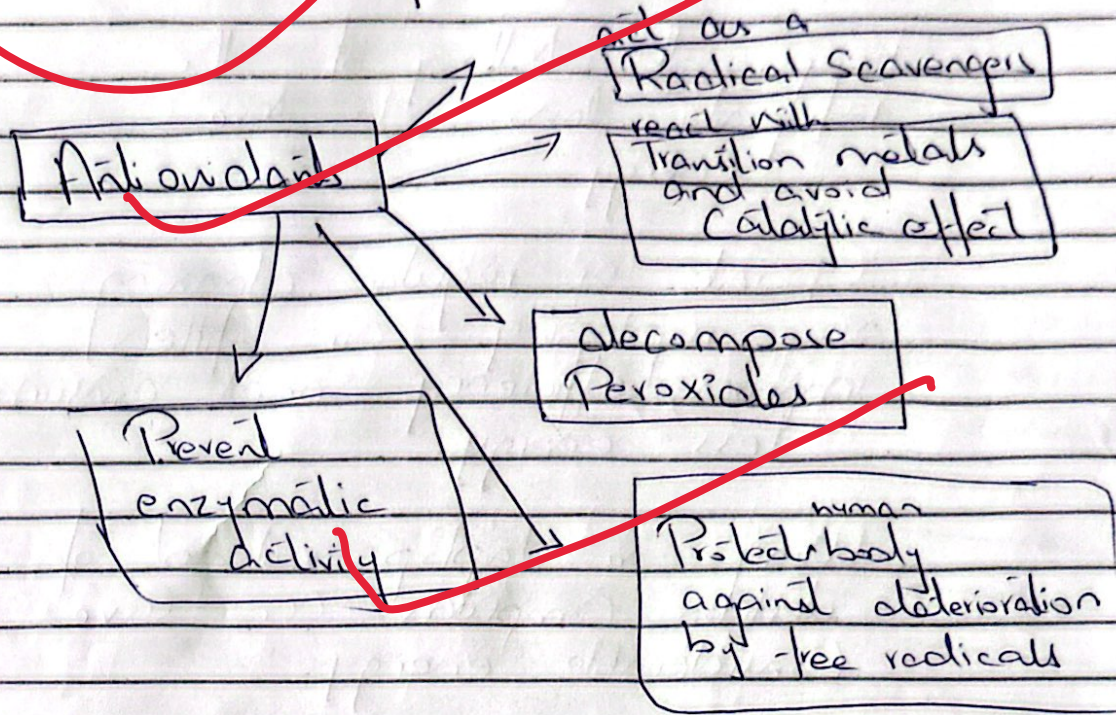
A large number of synthetic antioxidants are available for the stabilisation of non-food materials such as plastics, rubber and polymers.

Natural Antioxidants:-

Plants are the natural source of supplying man with valuable bioactive substances and thus

different plant products are being evaluated as natural antioxidants to improve the overall quality of meat and meat products.

Actions of Antioxidants



2
Qb Enlist few measures for energy conservation and its sustainable use.

Energy Conservation:

Energy Conservation is the act of reducing the waste and wastage of energy.

Measures for energy Conservation

Purchase devices and appliances which consume less energy.

Adapt Small Power strips.

Keep the setting of the refrigerator low to save energy.

Operate or regular cleaning or replacing air filters improves efficiency and consumes less energy.

Using a laptop instead of desktop computers can save considerable energy.

Cycling is the best way to save fuel.

Walking instead of driving also saves energy.

Skip the dryer on a breezy day.

Sustainable Uses

1. Save the cost and

Dont leave spaces after every line. Work on presentation

and lowers your utility bills.

2 Prolongs the existence of
↓ fossil fuels.

3 Protects the environment.

4 Reduces pollution

3

Explain complex concepts in simple terms.

Use real-life examples.

Include diagrams and flowcharts for competitive edge.

Discuss practical applications of scientific concepts.

Show all steps and working for calculations.

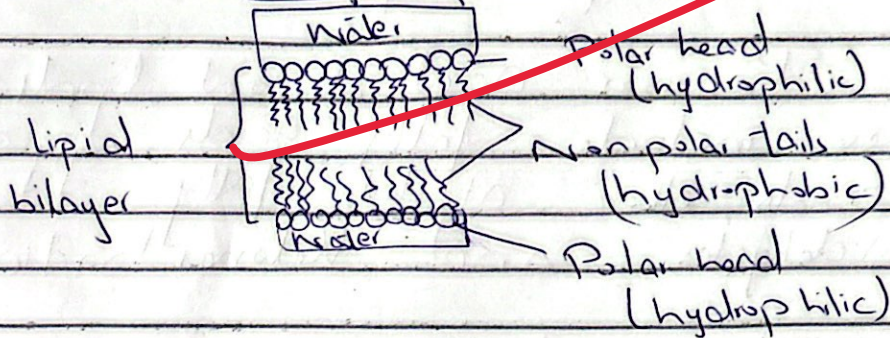
Use diagrams and graphs

Q No 2 Briefly explain lipids. What are some major types? What are their functions?

Lipids:

Lipids are organic compounds that contain oxygen, hydrogen and carbon and are the main constituent of the cell. Lipids are non-polar and are soluble in non-polar solvents but are insoluble in water as water is the polar and unable to dissolve non-polar.

⇒ Structure of lipid:



Types of lipid:

There are numerous specific types of lipids.

Simple lipids

Complex lipids

Precursor and derived lipids

Fatty Acids. (saturated and unsaturated - fatty acids.)

* Simple lipids:

Esters of fatty acids

with various alcohols.

1 -fats: Esters of fatty acids with glycerol

2 Waxes: Esters of fatty acids with higher molecular weight monohydric alcohols

* Complex lipids:

Esters of fatty acids

Containing groups in addition to alcohol and fatty acids.

Such as Phospholipids, Glycolipids, other complex lipids.

* Precursor and derived lipids.

These include fatty acids, glycerol, sterols, other alcohols, fatty aldehydes and ketone bodies.

* Fatty acids.

Fatty acids are carboxylic acids (or organic acids), usually with long aliphatic tails

Functions of lipids:

Waxes are found almost everywhere - the fruits and leaves of many plants possess many coatings, that

can safeguard them from small predators and dehydration.

Phospholipids:

Membranes are primarily composed of phospholipids.

Cholesterol:

Cholesterol is an important component of cell membranes and is also the basis for the synthesis of other steroids, including the sex hormones estradiol and testosterone as well as other steroids such as cortisone and vitamin D.

What is Hydrogen Bonding? Give elaborating structures and examples

Hydrogen Bonding:

Hydrogen bonding is based on dipole-dipole attraction between molecules of compounds.

Hydrogen Bonding is a bonding of hydrogen atom which is a partial positive with the electronegative atoms.

Conditions for Hydrogen Bonding:

More electronegativity:

More the electronegativity, more the strong hydrogen bonding.

Small size:

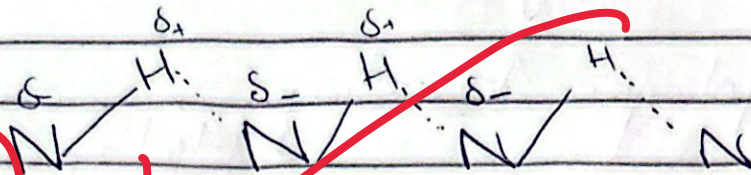
Small the size of the electronegative atom, more will be the good bond.

⇒ Structures elaborating Hydrogen Bonding:

Ammonia:

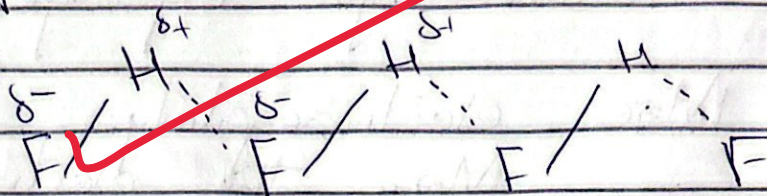
Hydrogen act as a partial positive δ^+ and Nitrogen act as a partial

negative δ^-



⇒ Fluorine act as most electronegative element

Fluorine being the most electronegative element bind very strongly with hydrogen through hydrogen bond



Q.No What is Nervous System.

Nervous System:-

The Nervous System is the controlling and regulating center of our body.

The Human nervous refers to the brain, nerves, spinal cord, ganglia and

Other receptor organs - that receive and interpret stimuli

Functions of the Nervous System:

The Nervous System Controls Brain growth and development.

Sensations (such as touch or hearing)

Movement, balance and coordination

Learning and memory

Healing and rehabilitation

Thought and emotion

Human Nervous System

