	Date://20 MON TUR WED THS FRI SAT	
	Assignment - 1	
	Mention the full qs statement for	
	proper evaluation. Addition that	
	these are just notes and cannot	
	be properly marks Carbohydrates are organic compounds, these	
	comprise of only carbon, hydrogen and Oxygen.	P
	Classification:	
	Combohydrates are charified into;	
(4)	Simple Combohydrates (monosaccharides, aligosaccharides)	
(2)	Complex Carbohydrates (Polysaccharides)	
	Monosaccharides	
	Monosaccharides are called simple sugar	-
	They are simplest sugar and cannot be hydroly	
<u> </u>	Their general formula is CH20)n. They are	-
	subdivided into trioses, etrose, pentose, hexoses et	C.
· · ·	Example	
	(1) Glucose: Immediate source of energy	
	(11) Galactoses A sugar in milk and yogurt.	
,	(iii) Fructose: A sugar in haney.	
	Oligosaccharides	×,
	Oligosaccharides are called compound sug	ers.
	On hydrolysis they yeild to 10 molecules	
5.5	of same or different manasaccharides, ??	
-		

Date:___/__/20_ Oligosaccharides youlding 2 molecules of monosa ccharides on hydrolysis is known as disaccharides. and the ones guilding 3 or 4 monasquehanides gre known as maccharide or tetraspechanide respectively and so on. Their general formula is Cn CH20 In-1. for disaccharide and Cn (HaO) n-2 for thisacchandles Example: (i) Sucrose: Common Aable sugar. Glucose + Fructose. (ii) Lactose: Major som in milk. rucose + Glactose. (iii) Maltose: Product of starch digestion. Glucose + Glucose. Draw the Kolysaccharides: Polysaccharides are complex sugars and re Heild were than 10 watermer of manosachands on hydrolysis. Hehropoly saccharides 66 Homopaly saccharides fello same type of mong saccharides on hydrolysis. 39 Examples Starch, glycogen, cellulose, pectin.

(8) Hetero poly saccharides Hetero polysaccharides yeild different types/ units of monosaccharides. 39 Example: Hyaluranic acid and Chandraitin. Characteristics of Carbahydrates. (1) Composed of monosaccharides that have a ring structure. (2) Monogaccharides are connected with glycosidic sprod. (3) Combohydrates one soluble in water. (4) Carbohydra, are sweet in taste. They are polyhydroxy aldehydes or ketones. (5) Simple carbohydrates digest quickly and (6) send immediate bursts of glucose (energy) into the bloodingoin. (7) Glucose is stored as glycogen in omimals and starch in plants. (8) Stored Combohydrate acts as energy source instead of postein. (9) Carbohydrates are rich in fiber content help to prevent constipution. They provide 3.9 calories of energy per gram. (10)

Flow Sheet:			
		1	
B	iamole cules		
			
Nucloic Add Co	rbohydrates	Protein lipid	
01	100		
Polysaccharin	I. Vlano sacchania	4	des
	Glucose	. szotlaM	
Homopolysarchanide	Heteropoly such oric	les	
Glycogen	Chondroiti		1
			100 ·
Structure:	НаОН ,		
	н	Glucose	
(1) Ho	Ho H	4 MCOSE	
	н он		
CH2OH CH2OH		UH20H	CHOH >
H H H	Hydrolysis	H H H	H
No on HO	- HOV	HO OH H	OH HOH
		н он	Н он
Glucose Gluc	.08 @	J	4
		glycos Bon	
		Maltose	
The state of the s			

PROTE INS Definition: Proteins are complex molecules made up of carbon, hydrogen, oxygen and nitrogen (sometimes sulphur and posphorus) Examples: Digestive enzyme, hemoglobin; Kerattn. Characteristics of Protein: 1. Proteins are polymers made up of monomers called the amino and. 2. Proteins carry out regulating activities as enzymes and harmones. Proteins one production as antibiotics. Proteins build we tissues of the body and 4. maintain and replace damaged tissues. Proteins are responsible for movement as 5. contractile proteins actin and myosin from basic structure of muscles. Keratin protein forms hair, nail, feathers, horns and beaks. Corrier protein ma molecules from one place to another around the body, eg. haemoglobin. Proteins provide 4.1 calories of energy per gram

01)

(1)



Classification of! Proteins

Classification of Protein based on

Structure of Proteins: There are four structural levels of organization

to describe the complex macrosolecules, protein based on the degree or complexity of the molecule.

They are primary secondary, Testimy and quaternary structure

Primary structure of Protein: The primary structure of protein exist as a long chain of amino acid arranged a particular

(Linear) sequence. The orie non-functional.

Secondary Structure of Protein: When the (2) polypeptides chain are coiled into a spiral

or helix to have on three dimensional Structure called secondary structure of

> protein, where the or mo acids interact by the formation of hydrogen bonds.

e.g; Keratin, silk fiber.

(3) Tertiony structure of proteins Long Polypeptide chains become more stabilizer by folding and coiling by the formation of lonic or hydrophobic bonds or disulphide bridges, this results in the tertiony structure of proton e.g. Globulins of blood. (4) Quaternary structure of Protein: When a protein is an assembly of more than one boly bebylge or enpired of 143 one this is said to the quaternary structure of stein. e.g. Hemoglobin and insulin. (02) Classification of Protein on the basis of Biological Function: Proteins can be classified on the basis of the biological function they perform (1) Enzymatic Protein: Ty are most varied and highly consulted proteins with contalytic activity. Frzymes caralyze a vanity of

D	Pate://20	
	reactions	
•	e.g. Urease, amylque, catalase etc.	
(2)	Structural Protein: These proteins helps to	
	protect biological structures.	
	e.g.; collagen, Keratin, collagen,	
3)	Transport Proteins transport or convier protein	
#	help in fromsport of ion or molecules in	
•	the body	
	e.g; Myoglobin, hemoglobin.	
		d
4)	Nutrient and storage Proteins: These provide	9
*)	nutrition to growing emblus and store	
	lans.	
	Flow Sheet:	
	Biological Molecule	9"
	Carbohydrates Protein Upid Nuclic Acid	
- tright, o		•
	(3) structural functional (2)	
	Protein Proteins	. •
	Primary Enzymatic	
Prot	ein + Secondary Structural Helping Fransport	Proteins.
	Quaternary Nutrient and Sharage	

	Date://20 LIPIDS OOOOO	
	(Fats +Oils)	
	Definition:	
	Lipids are naturally occurring organic	
	compounds, commonly known as oils and fats?	
1	- Triggeride , It is the basic unit of lipide	
	synthesized from glycerol so fatty acid.	
	- Structure:	
	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	
	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	
-	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	
	3 Fatty Acids Glycerol	
	Triglyceride	
*	- The second most common class of lipids.	
	are phospholipids. They contain glycerol and fatty	
	auds also contain physicaric and and low	
	molecular weight according	
	- Common phospholipids are lecithins and cephalins	
-	Classification of Lipids:	
	Lipids can be classified into following;	
-	(1) Saturated fat	
	(2) Trams fat	
	(3) Unsaturated fat	

Date:___/__/20. (1) Saturated Fat Saturated fats are solid at room temperature that's why they are also known as solid fat? Saturated fats can rise cholesteral. A healthy diet has less the 10% of daily calories from saturated fat. Animal foods. Scienated fats are mostly in animal foods such as milk, cheese and meat Poultry foods: They have relatively less saturated fat than red meat. Tropical Oils: They one also in tropical oils, such as excount oil, palm oil and coca butter. Others: Other sources mich contain saturated fats include; (snakes) snacks, non-dairy foods, coffee creamers, whipped toppings, butter, margarine, shortening etc. Trans Fat: (2) This is a fat that has been changed by a process of hydrogeration. This process increases the cheff life of fat and makes the for honder at room temperature. Trams fats also increase your chalesterol

	Date://20 Mon tue wed this fire sat	
	Level.	
	Sources:	
	Sources of Trans fats are as following;	
,	(1) Processed foods	
	(2) Snacks foods, such as chips and charkers	j.
	(3) Cookies	
	(4) Some margarine and salad dressings,	
	(6) foods made with shortening and	
	partially hydrogenated oils.	· ·
400		
(3)	Unsaturated Fats:	
	Unsaturated fats one liquid at room	
	temperature.	
	It is mostly oils from o'ms.	•
•	Unsaturated fats bys to improve the	•
	There are two types of unsaturated fats (1) Monounsaturated fat	
	(2) Poly unsaturated fout	
(1)	Monounsaturated fat:	
	Monounsaturated fats Keep good HDL	
	cholesteral Levels high.	
	sources: Avocado nuts, vegetable oils,	
	canola, olive, peanut oil.	

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(Ջ)	Polyunsaturated fat:
	Polyunsaturated fat in place of saturated
	fat may lower LDL chalesteral.
	This type of fat is mount in regentrables, will,
	and seafood.
	Omega-3 mo omega-6 fatty oulds one
	two types of polyunsaturated fat.
•	
A >	Characteristics of Lipids:
<u>1)</u>	Lipids are unsoluble in water.
2) (3)	They are soluble in organic solvents; ether.
<u>3)</u>	They have high energy content.
(<u>5)</u>	They act as an eletwical insulater
(6)	Saturated fait are solid at room temperature. Unsaturated fait is are liquid at room temperature.
(1)	Melting point of fat's depends on the length:
6	of the chain of constituent fatty and
	and degree of unsaturation.
(8)	They act as cellula metabolic regulators.
(9)	They protect many vital organs of
	body.
(10)	They help to regulate the body temperature.

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Good structure and presentation

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	Flow Sheet:	/
	Biomolecules	
-		
	Caxbohydrates Lipids Proton Nucleic aux	
	Trains Unsaturated Saturated	
	Monounsaturated Poly saturated	
	Omega - 6 Omega - 3	
	Description:	
-	Lipids: Organic compounds, fats and oil	
	Trans: Fromsitional shage (fatty acid) fats;	Snack
	Saturated: Solld fats; milk, meat, cheese	
	Unsaturated: liquid fats; regetable oil	
	Monunsaturated: found in puts, olive, palm oil	*
2.2	Polyunsatured so flower, sunflower.	
	Omage - 6: Carlol role in begin development	lama

Omega-3: Healthy fors; that support heart health