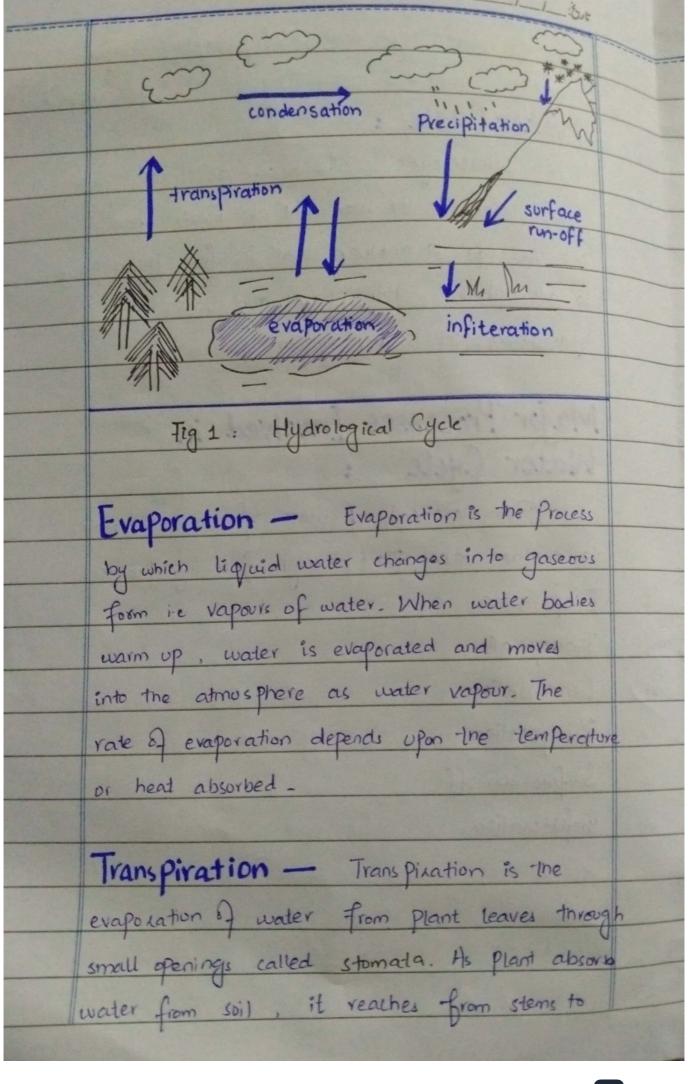
	What are different layers of atmosphere. On		
	what basis these layers are classified. In		
	which layers Auroras are.	formed and where do	
	satellites orbit - (CSS-		
	Atmosphere:	-	
	Atmosphere is the thin	blanket of gases	
	that surround the planet e		
	include		
	Fixed Gases	Variable 4ases	
	Nitrogen - 78%	Carbondioxide -0.03/	
	0xy gen - 21%	Ozone	
	gnert gases - 0.74%	Water vapours	
	LAYORS Of Atmosphe	re:	
	Layers of Almosphere:  Scientists have classified atmosphere into		
	o www. lourse based	on temperature	
	five distinct layers based		
	variations. These Layers	are ;	
i-	Troposphere	F - 1 CH   C   C   C   C   C   C   C   C   C	
ij -	Stratosphere	ment to the state of the state	
jii-	Mesosphere		
-			

iv-	Thermosphere	
V-	Exosphere	
	Exosphere	-
	1	
	Thermosphere	
	8° Mesopouse	
	Mesos Phere	
	Strato Pause	
4 6 3 4	Attitude	
	kn 40	
	30 Stratusphere	
	Tropo Pause	
	Troposphere Troposphere	
16-19	-100 -80 -60 -40 -20 0 20 40 60 80	
	Temperature °C	
3-23	9150 NE 100 183	
	Fig 1: Layers of Atmosphere	
i-	Troposphere:	
	The first layer of atmosphere is	
Bernell.	troposphere - It extends from the surface of	
	earth to 12 km above surface. This is the	
1.7.5.7	layer of atmosphere that supports life and	
	in a within	
	weather patterns. Commercial jets also fly within	
	this layer. As we move up from troposphere	
Barrier B.	Control of the Contro	

	, the temperature and the air Pressure decreases.
	The decrease in temperature is upto -53°c.
ii -	Stratosphere:
	Stratosphere is the second layer of atmosphere.
	It extends from 12-50 km above the surface
	of earth. It is within this layer that
	Ozone layer exists. This layer Protects the
	earth from incoming ultraviolet radiations, which
	otherwise are hamful for life on earth. It is
	because of Prescence of this layer and absorption
	of UV radiations, the temperature here
	increases dramatically up to -3°c.
	Mesosphere:
	Mesosphere is the middle layer of
	atmosphere. It extends from 50-80 km above
	The surface of earth. Meteous are destroyed
1	within this layer. This layers absorbs no heat
	and is the coldest layer of atmosphere. The
	temperature declines to -73°C.
	Thermosphere:
	Thermosphere is the hottest layer of
	atmosphere. It extends from 80-700 km.
	It directly absorbs incoming radiations from

the sun- It is within this layer aurolas are formed that are observed in the sky of Polar regions. The temperature here exceeds 2000°C Exosphere: The outer most layer of atmosphere is exosphere. It extends from 100 to 10,000 km and blends into outer space. The temperature here varies. It is within this layer that satellites and space crafts are found-Elect sateletties + Exosphere spare crats 100,000 km Aurora Thermos Phere (80-700 km) Meteors Mesos Phere (80-80 km) 03 03 03 03 Layer Stratosphere (12-50 km) Weather Commerciallets Troposphere (0-12 Km) Figa: Layers of atmosphere composition/function

	Describe water cycle and briefly explain major
	Processes involved in water cycle? (CSS-2021)
	Water Cycle:
	Mater cycle, also known as hydrological
	cycle, refers to the movement of water from
	one reservoir to another that is from hydrosphere
	to atmosphere, lithosphere and back to
	hydrosphere. It is a cyclical process.
	Major Processes Involved in
	Water Cycle:
	The movement of water from one reservoir
	to another takes place through following Processes:
i-	Evaporation
ii -	Transpiration
)ii -	Condensation
iv-	Precipitation
	Surface run-off
V-	
vi-	Infilteration
	The state of the s
	The second secon



leaves and from there excessive water is removed via transpiration. Transpiration adds more water into the atmosphere. Condensation - Condensation is the opposite of evaporation. It is the process in which water vapours change into liquid displets. The small dioplets are suspended into the atmosphere in the form of clouds on fog. Precipitation - Precipitation refers to the fall of water particles from atmosphere to ground in the form of Rain, snow, sleet or hailstorm. Surface Run-off - Much of the water that falls to the ground runs-off into streams which later join rivers and finally reach to ocean, where a great deal of evaporation occurs. The run-off also carries nutrients and sediments along with it -

Infilter	ration —	4 of Meration	refers to
the Process	whereby grou	nd soaks -	he rainwater
theough s	oil and Pozous	rocks. This	helps in
vechaiging	ground water	level-	
	Andreas State Contract		sanahar)
Importa	nce of Wa	ter Lycle	:
Mater cycl	e has a numb	per of func	tions listed
as;	GI LAKASIK	SVA DISINEN	bellow of
->	purifies surface	water by	evaporation
->	sustains life by	y Providina	fresh water
-> H	nelps in biogeoc	hemical cyc	les
->	recharges grow	und water -	appurfers
	maintains ter	Perature o	and climate
	Treat terms	-	
		-	
SERVICE DE	- Asiansian .	- 4,00	Was and the
	0 007 77 6	LY busin	354
			1000
	4		
30 A 10 A	6 To 1630		THE RESERVE

What is acid rain. Describe its causes and how it can be prevented? ( CSS-2018) Acid Rain: Acid rain refers to rain or any other form of Precipitation that is unusually acidic ite it has high level of hydrogen ions. The PH of aid main is less than 5.6. Acid Rain Formation: Explaination Acid rain is produced when gases such as possides of sulfur (SOx) and oxides of nitrogen (NDx) are released into the atmosphere. These gases react with unter vapours in the Prescence of sunlight and from aidic particles including sulphusic acid (HISOu) and nitric acid (HNO3). These Particles fall to ground either as wet deposition ( Lain, snow, hail) and dry deposition ( dust and smake Particles Acidic Yases & Their Sources:

Carbondioxide  Burning of fessil fiels  Industries, respiration  Methane  Landfils, wetlands,  animals, Paddy fields  Oxides of Sufor  Volcanoes, Ocean,  burning of fossil fiels  industries  SOX+ HOD HESOY  NOX+ HOD HISOY			T - 1- 30
Methane Landfils, wellands, animals, Paddy fields  Orides of Sufur Volcanoes, Ocean, burning of fostil foels  Orides of Nitrogen Venicles, fossil foels  industries  SOX+ HaD HASOY  NOX+ HAD	Car	bondioxide	Burning of fossil Cal
Methane Landfils, wetlands,  animals, Paddy fields  Oxides of Sulfor Volcanoes, Ocean,  burning of fossil foels  Venicles, fossil foels  industries  SOX+ HOD HISOY  NOX+ HOD HISOY  NOX+ HOD HINOS  IN Wet Deposition  Fig 1: Acid Rain  Effects of Acid Rain:  Drid vain can have damaging effect			Endustries, respiration
Orides of Sufor Volcanoes, Ocean,  burning of fosil foels  Venicles, fossil foels  industries  SOX+ HAD HASOY  NOX+ HAD HASOY  NOX+ HAD HASOY  NOX+ HAD HASOY  Wet Deposition  Ty Deposition  Ty Deposition  Fig 1: Acid Rain  Effects of Acid Rain:  Did vain can have damaging effect	Me	thane	Landfills, wellands
Duides of Sulfor Volcanoes, Ocean,  burning of fosil feels  Duides of Nitrogen Venicles, fossil feels  industries  SOX+ HAD HASOY  NOX+ HAD HA			animals, Paddy field
Drudes of Nitrogen Venicles, fossil foels industries  SOX+ H20 H2504 NOX+ H20 HNO3 NOX+ H20 H2504 NOX+	Orides	of Sufor	Volcanoes, Ocean,
Sox + H20 H2504  Sox + H20 H2504  NOX + H20 HNO3  NOX + H20 HNO3  Wet Deposition  Tig 1: Acid Rain  Effects of Acid Rain:  Puid vain can have damaging effect		-	burning of fosil fock
SOX+ H20 H2SOY NOX+ H2O HNO3  NOX+ H2O HNO3  Wet Deposition  Tig 1: Acid Rain  Effects of Acid Rain:  Acid Rain:	Oxides	of Nitrogen	Venicles, fossil foels
Tig 1: Acid Rain  Effects of Acid Rain  Phid rain can have damaging effect		, 0	
Fig 1: Acid Rain  Effects of Acid Rain  Phid rain can have damaging effect			
Fig 1: Acid Rain  Effects of Acid Rain  Phid vain can have damaging effect			
Tig 1: Acid Rain  Effects of Acid Rain  Phid vain can have damaging effect	- Farm	(n. + H)	H2SOy **
Fig 1: Acid Rain  Effects of Acid Rain:  Pid vain can have damaging effect	1	NOX + HLO	HNO3
Fig 1: Acid Rain  Effects of Acid Rain:  Pid vain can have damaging effect	[1	Boile	3 may 11 1/1/
Fig 1: Acid Rain  Effects of Acid Rain:  Pid vain can have damaging effect	43	Dry	Deposition , , , Wet Deposition
Effects of Acid Rain:  Acid vain can have damaging effect			
Effects of Acid Rain:  Acid vain can have damaging effect	440	Change 1	
Effects of Acid Rain:  Acid vain can have damaging effect	5.5		The second secon
Effects of Acid Rain:  Acid vain can have damaging effect	-10-	-00	
Effects of Acid Rain: Acid vain can have damaging effect	Ŧŧ	91: A	Icid Rain
Acid vain can have damaging effect			
Acid vain can have damaging effect	Ecc	C A 24 1	Patra
both on human health as well as environment.	Effects 6	Hua	Acoust 2
both on human health as well as environment.	Più	d vain can	have damaging effect
	noth on 1	numan hea	Ith as well as environment.

Environment	
Water Bodies and Aquatic Life - Aud	-
turns water bodies (lakes, streams) audic. At	
low PH most p fish eggs donot hatch and many	-
adult fish (trout) cannot survive an acidic water	
Acid rain thus reduces aquatic biodirecity.	
a- Soil - Acid rain damages soil structure and	
composition. At also affects soil nutrient holding	
capacity. Many soil microorganisms cannot survive	
in low PH.	
3- Damages Vegetation - Acid rain damages forests	
by causing leaves to fall, Storrled growth and	
damaged back. The Pollutants also damage stomatal	
opening thus plants cannot take oxygen properly.	
Human Health	
Acid is rain water is diluted and thus cause	
no direct have to human health However; the	
causative agents of acid rain such as SOx, Nox	-
can have significant impact on human causing	-
cardiac and vespiratory problems such as asthama	-1

branchitis, and damage to skin. Historical Buildings & Monuments: Acid rain can also alamage historical buildings and monuments. The acidic components react with marble and lime and form founder like substance called gypsum that fakes away. H2504 + CaCO3 -> CaSO4 + H2CO3 Many historical monuments such as Taj Mahl in India and Statue of liberty have been damaged owing to acid rain -Prevention of Acid Rain: Acid rain can be prevented by taking significant measures to reduce emissions of SDx and NOx into atmosphere. The preventive measures can range from source control, use of Pollution control technologies (PCTs) to shifting to alternate energy sources. A few Preventive measures are listed ous follows: Use of Alternate Energy - Alternate energy resources such as renewables (solar, hydal, wind)

	and brofoels can help a great deal in cutting
STATE OF THE PARTY	mission of harmful gases responsible for and rain
in	actmos Pheve -
2- 8	Better Quality of fuel - Instead of
U	sing low quality fuel with high sulfor content
	shifting to better quality fuel with low
	ulfor content is adviceable - Similarly other
	easures such as coal washing, wet scrubber
	extrostatic Precipitators can also reduce emissions.
	atural gas is a better solution compared to
	ssil fuels since it produces less emissions.
1	SMI PES SINCE I
. (0	italytic Converters - Use of catalytic
3-	nverters in automobiles is another feasible
LO	h 1 D values emissions The use of metal
m	nethod of reducing emissions. The use of metal
wa:	ted beads converts harmful gases into less
hav	rmful gases.
	1 10
4- Lir	ning - Liming of soil also helps in
red	veing the damaging effect of aid rain on
soil	- However; this method is feasible only
for	a small scale use.
1	

Policy Guidelines - The formation of appropriate Policies to control air Pollution and emissions within a certain level also helps in curtailing the harmful effects of pollutants and their aftermatns. Public Awareness - Educating the masses and creating ownerers among general Public Regarding the harmful consequences of their careless actions can also help a great deal. Common evo-friendly Practices include; Use of Public transport > Carpooling -> Retiance of on renewables (solar /wind) -> Plantation